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## Structure Reports

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## Dong-Dong Lin, ${ }^{\text {a }}$ Kai-Liang Yin ${ }^{\text {b }}$ and Duan-Jun Xu ${ }^{\text {a }}$

${ }^{\text {a }}$ Department of Chemistry, Zhejiang University, Hangzhou 310027, People's Republic of China, and ${ }^{\mathbf{b}}$ Key Laboratory of Fine Chemical
Engineering, Jiangsu Polytechnic University,
Changzhou 213016, People's Republic of China
Correspondence e-mail: xudj@mail.hz.zj.cn

## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.027$
$w R$ factor $=0.071$
Data-to-parameter ratio $=13.1$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## catena-Poly[[[bis(1 H -imidazole- $\kappa \mathrm{N}^{3}$ )zinc(II)]-$\mu$-terephthalato- $\left.\kappa^{2} O: O^{\prime}\right]$ trihydrate]

In the title polymeric complex, $\left\{\left[\mathrm{Zn}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{~N}_{2}\right)_{2}\right]\right.$-$\left.3 \mathrm{H}_{2} \mathrm{O}\right\}_{n}$, the $\mathrm{Zn}^{\mathrm{II}}$ atom is coordinated by two terephthalate dianions and two imidazole ligands in a distorted tetrahedral geometry. The terephthalate dianions are located on inversion centres and bridge neighbouring $\mathrm{Zn}^{\mathrm{II}}$ atoms to form zigzag polymeric complex chains. The interplanar distance of 3.212 (17) $\AA$ between parallel imidazole rings of adjacent polymeric chains suggests the existence of $\pi-\pi$ stacking.

## Comment

As $\pi-\pi$ stacking between aromatic rings is correlated with the electron-transfer process in some biological systems (Deisenhofer \& Michel, 1989), we are interested in the study of the nature of $\pi-\pi$ stacking in metal complexes (Liu et al., 2004; Pan \& Xu, 2004). As a part of our ongoing investigations, the title polymeric $\mathrm{Zn}^{\mathrm{II}}$ complex, (I), incorporating imidazole, has recently been prepared, and its X-ray structure shows the existence of $\pi-\pi$ stacking between imidazole rings.

(I)

The structure of a fragment of (I) is shown in Fig. 1. The $\mathrm{Zn}^{\mathrm{II}}$ atom is coordinated by two imidazole ligands and two terephthalate dianions in a distorted tetrahedral geometry. The bond angles involving the metal centre range from 103.64 (7) to 122.75 (8) ${ }^{\circ}$ (Table 1). The terephthalate dianions are located on inversion centres and bridge neighbouring $\mathrm{Zn}^{\text {II }}$ atoms through terminal carboxyl groups to form zigzag polymeric complex chains (Figs. 1 and 2). The carboxyl groups of the terephthalate ions coordinate to the $\mathrm{Zn}^{\mathrm{II}}$ atom in a monodentate manner. The uncoordinated carboxyl O atoms (O2 and O 4 ) are hydrogen-bonded to the uncoordinated water molecules, resulting in a linkage between adjacent polymeric chains (Fig. 1).

The interplanar distance of 3.212 (17) $\AA$ between parallel imidazole rings suggests the existence of $\pi-\pi$ stacking (Fig. 2).

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Figure 1
The structure of a fragment of (I), showing $30 \%$ probability displacement ellipsoids [symmetry codes: (v) $2-x,-y,-z$; (vi) $1-x, 2-y, 2-z$ ]. Hydrogen bonds are shown as dashed lines.

## Experimental

All reagents were commercially available and of analytical grade. An ethanol solution ( 5 ml ) of imidazole ( 2 mmol ) was mixed with an aqueous solution ( 5 ml ) of $\mathrm{ZnCl}_{2}(1 \mathrm{mmol})$, and the mixture was refluxed for 1 h . An aqueous solution ( 8 ml ) containing terephthalic acid ( 1 mmol ) and $\mathrm{Na}_{2} \mathrm{CO}_{3}(2 \mathrm{mmol})$ was then added to this mixture, which was refluxed for a further 1 h . After cooling to room temperature, the solution was filtered. Colourless single crystals of (I) were obtained after 20 d .

## Crystal data

$\left[\mathrm{Zn}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{3} \mathrm{H}_{4} \mathrm{~N}_{2}\right)_{2}\right] \cdot 3 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=419.69$
Triclinic, $P \overline{1}$
$a=8.3046(4) \AA$
$b=9.5443(4) \AA$
$c=12.0048(8) \AA$
$\alpha=81.220(3)^{\circ}$
$\beta=70.030(2)^{\circ}$
$\gamma=83.678(2)^{\circ}$
$V=882.10(8) \AA^{\circ}$
Data collection

| Rigaku R-AXIS RAPID | 3076 independent reflections |
| :--- | :--- |
| $\quad$ diffractometer | 2807 reflections with $I>2 \sigma(I)$ |
| $\omega$ scans | $R_{\text {int }}=0.024$ |
| Absorption correction: multi-scan | $\theta_{\max }=25.0^{\circ}$ |
| $\quad(A B S C O R ;$ Higashi, 1995) | $h=-9 \rightarrow 9$ |
| $T_{\min }=0.665, T_{\max }=0.860$ | $k=-11 \rightarrow 11$ |
| 6595 measured reflections | $l=-14 \rightarrow 14$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.027$
$w R\left(F^{2}\right)=0.071$
$S=1.12$
3076 reflections
235 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0328 P)^{2}\right. \\
& +0.4577 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3 \\
& (\Delta / \sigma)_{\text {max }}=0.001 \\
& \Delta \rho_{\text {max }}=0.23 \mathrm{e}^{-3} \\
& \Delta \rho_{\text {min }}=-0.35 \mathrm{e}^{-3} \\
& \text { Extinction correction: none }
\end{aligned}
$$



Figure 2
The partially overlapped arrangement of imidazole rings, showing $\pi-\pi$ stacking [symmetry code: (i) $1-x, 1-y, 1-z$ ]. Hydrogen bonds are shown as dashed lines.

Table 1
Selected geometric parameters $\left(\AA^{\circ},^{\circ}\right)$.

| $\mathrm{Zn}-\mathrm{O} 1$ | $1.9883(15)$ | $\mathrm{Zn}-\mathrm{N} 1$ | $1.9735(18)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Zn}-\mathrm{O} 3$ | $1.9660(15)$ | $\mathrm{Zn}-\mathrm{N} 3$ | $1.9940(19)$ |
|  |  |  |  |
| $\mathrm{O} 1-\mathrm{Zn}-\mathrm{O} 3$ | $103.95(6)$ | $\mathrm{O} 3-\mathrm{Zn}-\mathrm{N} 1$ | $106.84(7)$ |
| $\mathrm{O} 1-\mathrm{Zn}-\mathrm{N} 1$ | $111.57(7)$ | $\mathrm{O} 3-\mathrm{Zn}-\mathrm{N} 3$ | $103.64(7)$ |
| $\mathrm{O} 1-\mathrm{Zn}-\mathrm{N} 3$ | $106.29(7)$ | $\mathrm{N} 1-\mathrm{Zn}-\mathrm{N} 3$ | $122.75(8)$ |

Table 2
Hydrogen-bond geometry ( $\AA,{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 1 W-\mathrm{H} 13 \cdots \mathrm{O} 2$ | 0.95 | 1.87 | 2.818 (2) | 174 |
| $\mathrm{O} 1 W-\mathrm{H} 14 \cdots \mathrm{O} 2 W^{\text {i }}$ | 0.87 | 1.91 | 2.768 (3) | 170 |
| $\mathrm{O} 2 W-\mathrm{H} 15 \cdots \mathrm{O} 4$ | 0.85 | 2.03 | 2.886 (3) | 177 |
| $\mathrm{O} 2 W-\mathrm{H} 16 \cdots \mathrm{O} 2^{\text {ii }}$ | 0.93 | 1.98 | 2.893 (3) | 168 |
| $\mathrm{O} 3 W-\mathrm{H} 17 \cdots \mathrm{O} 1 W^{\text {ii }}$ | 0.88 | 1.94 | 2.790 (3) | 164 |
| $\mathrm{O} 3 W-\mathrm{H} 18 \cdots \mathrm{O} 4^{\text {iii }}$ | 0.90 | 1.95 | 2.821 (3) | 163 |
| $\mathrm{N} 2-\mathrm{H} 6 \cdots \mathrm{O} 1 W^{\text {iv }}$ | 0.86 | 2.03 | 2.877 (3) | 168 |
| $\mathrm{N} 4-\mathrm{H} 10 \cdots \mathrm{O} 3 \mathrm{~W}$ | 0.86 | 1.94 | 2.781 (3) | 164 |

Symmetry codes: (i) $-x+1,-y+1,-z+1$; (ii) $x, y+1, z$; (iii) $x+1, y, z$; (iv) $x-1, y, z$.

H atoms on aromatic rings were placed in calculated positions, with $\mathrm{C}-\mathrm{H}=0.96 \AA$ and $\mathrm{N}-\mathrm{H}=0.86 \AA$, and were included in the final cycle of refinement in the riding model, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}$ of the carrier atoms. Water H atoms were located in a difference Fourier map and refined as riding in their as-found positions relative to the O atoms, with fixed isotropic displacement parameters of $0.05 \AA^{2}$.

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/ MSC, 2002); program(s) used to solve structure: SIR92 (Altomare et al., 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP3 for Windows (Farrugia, 1997) and $X P$ (Siemens, 1994); software used to prepare material for publication: $\operatorname{Win} G X$ (Farrugia, 1999).

## metal-organic papers

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